

This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1915 Excerpt: ...salt. The substance melted at 2130 and had a specific rotation of 16.85. No corresponding salt of chitonic acid has been described. Discussion of Results.--The results here enumerated permit only of the conclusion that chondrosamine is isomeric with glucosamine. The fact that in the process of the osazone formation the amino group is eliminated from the molecule, renders it probable that the group is attached to the α -carbon (to the carbonyl group). The details of the structure of chondrosamine will be established by further investigation. Work in this direction is now in progress. In conclusion we wish to refer again to an observation made in Hofmeisters laboratory. Kondo³ prepared on treatment of the 2 Ber. d. deutsch. chem. Gesellsch., xxvii, p. 140, 1894. 3 Biochem. Zeitschr., xxvi, p. 116, 1910. products of hydrolysis of chondroitin sulphuric acid an osazone having a melting point at 143 C. Discussing this observation Kondo referred to the similarity of the osazone with that of xylose. We inferred that the discussion indicated the authors inclination to regard xylose as a probable constituent of chondroitin sulphuric acid. Since then Professor Hofmeister informed us in private correspondence that both he and his co-workers were convinced that the similarity of their osazone with xylosazone was only superficial, and that we attached undue emphasis to their discussion. At present it seems to us probable that their osazone was that of chondrosamine, but in a state of imperfect purity. EXPERIMENTAL. Preparation of Hexosamine Hydrochloride. Seventy-five grams of chondroitin sulphuric acid barium salt were hydrolyzed for about seven and a half hours with 400 cc. of 20 per cent hydrochloric acid, with the addition of 15 grams of stannous chloride. Barium su...

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